

## Nuclear Decay Work Sheet Answer Key

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### Nuclear Decay Work Sheet Answer

Introduction to Nuclear Chemistry — Page Name: Class/Lab Period: Nuclear Equations Worksheet  
Identify the missing atomic nuclei or radiation particles in the following nuclear equations: 1. Alpha decay of radium-226, the most abundant isotope of radium  $^{226}\text{Ra} + ^4_2\text{He}$  2. Radioactive decay of carbon-14, which is used in radiocarbon dating 3.

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### **NUCLEAR DECAY Predict the products of the following ...**

Nuclear Decay. The following atoms all undergo alpha particle emission. Write the complete nuclear equation.  $\rightarrow$  alpha particle + Pb-206  $\rightarrow$  alpha particle + Th-234  $\rightarrow$  alpha particle + Ra-234  $\rightarrow$  alpha particle + Po-218. The following atoms all undergo beta decay. Write the complete nuclear equation.  $\rightarrow$  beta particle (e-) + N-14  $\rightarrow$

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Radioactive Dating, Radioactive Tracers (both medical and agricultural uses), Radiation Therapy (chemotherapy) V. Half-life - is the time required for half the atoms of a radioactive nuclide to

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decay. Each radioactive nuclide has its own half-life. Example: Carbon-14 has a half-life of 5715-5730 years. Page 708

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